ORIGINAL ARTICLE

Simultaneous determination of amoxicillin and potassium clavulanate in combined medicinal forms: procedure transfer from HPLC to UPLC

Současné stanovení amoxicilinu a klavulanátu draselného v kombinovaných lékových formách: přenos postupu z HPLC na UPLC

Anna O. Dobrova • Olga S. Golovchenko • Ivan V. Bezruk • Liudas Ivanauskas • Victoriya Georgiyants

Received May 26, 2020 / Accepted September 18, 2020

Summary

This article presents the results of transferring a highperformance liquid chromatography (HPLC) procedure for the assay of amoxicillin and potassium clavulanate in tablets to the ultra-performed liquid chromatography (UPLC) conditions. Since the State Pharmacopoeia of Ukraine (SPhU) does not contain the monograph for the simultaneous analysis of amoxicillin and clavulanic acid, the British Pharmacopoeia procedure was used. Parameters of the procedure were optimized to fit the UPLC and to make a better performance. Transfer of the method to the UPLC conditions allowed to shorten the run time from 15 min to 7.5 min, which makes the process less time-consuming and more cost-effective. The upgraded procedure was further validated. Validation of both methods was performed in terms of linearity, precision, accuracy, specificity and stability. HPLC method was verified to later implementation into the SPhU's monograph. Afterwards, the methods were compared in terms of their impact on the environment using the eco-scale that included hazards of the solution, the amount of produced wastes, the impact on environmental and laboratory staff, etc. Both methods appeared to be eco-friendly with a moderate advantage of UPLC method. Moreover, the statistical comparison

was performed using Passing Bablok regression method. It showed that both methods are statistically comparable. **Key words:** amoxicillin/clavulanate formulations • UPLC • HPLC • method comparison • green chemistry

Souhrn

Článek představuje výsledky přenosu postupu vysoce účinné kapalinové chromatografie (HPLC) pro stanovení amoxicilinu a klavulanátu draselného v tabletách do podmínek ultravysokoúčinné kapalinové chromatografie (UPLC). Protože Státní lékopis Ukrajiny (SPhU) neobsahuje monografii pro simultánní analýzu amoxicilinu a kyseliny klavulanové, byl použitý postup podle Britského lékopisu. Parametry postupu byly optimalizovány tak, aby vyhovovaly UPLC a zlepšily výkon. Přenos metody do podmínek UPLC umožnil zkrátit dobu chodu z 15 minut na 7,5 minuty, což činí proces méně časově náročným a nákladově efektivnějším. Vylepšený postup byl dále ověřen. Validace obou metod byla provedena z hlediska linearity, přesnosti, správnosti, specificity a stability. Metoda HPLC byla ověřena pro pozdější implementaci do monografie SPhU. Poté byly metody porovnány z hlediska jejich dopadu na životní prostředí pomocí ekologické stupnice, která zahrnovala rizika řešení, množství vyprodukovaných odpadů, dopad na pracovníky životního prostředí a laboratoře atd. Obě metody se ukázaly být ekologické s mírnou výhodou pro metodu UPLC. Statistické srovnání bylo navíc provedeno regresní metodou Passing Bablok. Ukázalo se, že obě metody jsou statisticky srovnatelné.

Klíčová slova: formulace amoxicilin/klavulanát • UPLC • HPLC • srovnání metod • zelená chemie

Anna O. Dobrova – PhD student ⊠) • O. S. Golovchenko • I. V. Bezruk • V. Georgiyants

Department of Pharmaceutical Chemistry National University of Pharmacy 61100 Valentynivska Str. 4, Kharkiv, Ukraine e-mail: anna.dobrova08@gmail.com

L. Ivanauskas

Department of Analytical and Toxicological Chemistry Lithuanian University of Health Sciences A. Mickevičiaus Str. 9, LT 44307 Kaunas, Lithuania

Introduction

Amoxicillin is a broad-spectrum semi-synthetic β -lactam antibiotic. Recently, it has been widely used

in combination with clavulanic acid or its derivative potassium clavulanate. Clavulanic acid is a β-lactamase inhibitor, which prevents the enzymatic destruction of the β-lactam ring of amoxicillin by β-lactamase, and has also shown minimal antibacterial activity¹⁾. According to the World Health Organisation's (WHO) latest list of Critically Important Antimicrobials for Human Medicine, β-lactam antibiotics with β -lactamase inhibitors are the high priority drugs since they have shown a lower resistance potential and have a high frequency of use for various indications. Thereby, it is recommended as a first-choice medicine for treating upper and lower respiratory tract diseases, hospital and community-acquired pneumonia, exacerbations of chronic obstructive pulmonary disease (COPD), etc. There is also a possibility of its usage in the treatment of complications of novel virus infections of the respiratory tract and to prevent secondary infection²⁻⁴⁾. Amoxicillin still plays a significant role in the treatment of Helicobacter pylori associated gastritis according to the last Maastricht V consensus since its moderate resistance⁵⁾. It is also used in the treatment of complicated intraabdominal infections, lower urinary tract infections, skin and soft tissue infections, etc. The chemical structures of amoxicillin and clavulanic acid are presented in Figure 1.

Effectiveness of the therapy depends on the quality of active pharmaceutical ingredients which are components of dosage forms. The State Pharmacopoeia of Ukraine does not have any requirements for quality control of medicines that contain amoxicillin and potassium clavulanate. Thus verification of the method is necessary to deliver qualitative treatment for patients in Ukraine.

European Pharmacopoeia as well as the United States Pharmacopoeia recommend estimating the quality of amoxicillin and potassium clavulanate combined medicinal forms by high-performance liquid chromatography (HPLC)^{6, 7)}. With technology improvement, ultra-performance liquid chromatography (UPLC) method provides some notable advantages such as speed, while maintaining high sensitivity and resolution comparable to HPLC. These advantages can be useful for manufacturers to reduce costs during analysis. The UPLC analysis is less time-consuming, which makes the analysts' time more efficient and productive.

The analytical method selection should not be only in terms of accuracy or precision but also its environmental and health safety. Whereas the UPLC method takes fewer volumes of solvent, it can be hypothetically assumed to be eco-friendlier and less harmful to analysts' health⁸).

In this study, we aimed to transfer HPLC procedures to UPLC to check the equality of measurements and compare their environmental and health impact.

Experimental part

Chemicals and reagents

Amoxicillin and potassium clavulanate standards were purchased from Sigma-Aldrich GmbH (Steinheim, Germany). All solvents were HPLC grade, methanol was purchased from Honeywell (Germany) and purified water was obtained from a Mili-Q purification system (Millipore, Burlington, MA, USA). Five manufacturing batches (184153, 187489, 184111, 185691, 184117) of combined amoxicillin and clavulanic acid tablets – AMOXIL-K625 (Arterium Corporation, Ukraine) were purchased in the local pharmacy and taken as study samples.

Equipment

Liquid chromatography separation was performed using a Shimadzu Nexera X2 LC-30AD HPLC system (Shimadzu, Japan) composed of a quaternary pump, an on-line degasser, a column temperature controller, a SIL-30AC autosampler (Shimadzu, Japan), a CTO-20AC thermostat (Shimadzu, Japan) as well as a SPD-M20A diode array detector (DAD). Other instruments such as an Ultrasonic Cleaner Set for ultra-sonication use (Wise Clean WUC-A06H, Witeg Labortechnik GmbH, Germany), a Libra UniBloc AUW120D (Shimadzu Analytical Scale, Japan); "A" class volumetric flasks that meet requirements of the State Pharmacopoeia of Ukraine were used in the investigation⁹⁾.

Stock and working solutions

Buffer solution pH 4.4. 7.8 g of sodium phosphate monobasic was dissolved in 950 ml of water R and adjusted to pH 4.4 \pm 0.05 with phosphoric acid diluter, then filled up to the volume of 1000.0 mL with the same solvent, and mixed.

The test solution. To 500 mg (accurate weight) of Amoxil-K625 tablets powder (equivalent to 250 mg of amoxicillin) 400 ml of water were added, shaken at a speed of 400 rpm (Shaker SM 30 B (Edmund Buhler, Germany)) for 20 minutes, and filled up to the volume of 500 mL.

Standard solution. 100 mg (accurate weight) of amoxicillin trihydrate analytical standard and 80 mg (accurate weight) of potassium clavulanate analytical

$$a$$
 NH_2
 H
 O
 OH
 A
 OH
 A

Fig. 1. The structural formula of amoxicillin (a) and potassium clavulanate (b)

standard were placed into a 200 mL volumetric flask, dissolved in water, filled up to the mark, and mixed.

Placebo solution. Accurate weight of placebo (500 mg) was placed in a volumetric flask of a volume of 500 mL, 400 mL of water were added, shaken at a speed of 400 rpm (Shaker SM 30 B; Edmund Buhler, Germany) for 20 minutes, filled up to the volume of 500 mL, and mixed.

All solutions were filtered through a membrane filter with a pore diameter of $0.45 \mu m$.

Chromatographic conditions for HPLC analysis

The separation of components was performed on a RP column ACE C18 (250 mm \times 4.6 mm, with particle size of 5 µm, Pennsylvania, USA). The composition of the mobile phase was buffer solution pH 4.4 and methanol in the ratio 95 : 5, respectively. The flow rate was set at 1.0 mL/min. The wavelength was set at 220 nm. 20 µL was injected into the chromatographic system.

Chromatographic conditions for UPLC method

The analysis was performed with an ACQUITY UPLC BEH C18 column (50 mm \times 2.1 mm, with particles size 1.7 μ m, Milford, USA). The composition of the mobile phase was buffer solution with pH 4.4 and methanol in the ratio 98 : 2, respectively. The detection was performed at 220 nm with a flow rate at 0.1 mL/min. The volume of injection was 0.2 μ L.

Method validation

Both HPLC and UPLC procedures were validated according to the requirements of ICH guidelines and the State Pharmacopoeia of Ukraine in terms of linearity, precision, accuracy, specificity, and stability^{9, 10)}.

Assessment of methods influence on the environment

Parameters of the analytical methods such as the number of reagents, hazards, energy, and waste define its impact on the environment^{11, 12)}. Determination of method greenness was performed using Eco-scale, where the ideal value is 100. If some parameters of analysis differ from the principles of green chemistry, penalty points are assigned. The sum of

penalty points obtained after revision is taken into account for Eco-scale calculation using the following formula:

- Analytical Eco-scale = 100 penalty points
- > 75 represents an excellent green method
- > 50 represents an acceptable green method
- < 50 represents an inadequate green method

Globally Harmonized System (GHS) of Classification and Labeling of Chemicals provides full information about the determination of safety class of reagent based on physical, environmental, and health hazards¹³. M. Tobiszewski et al. proposes determination of penalty points for each reagent by multiplying the amount of GHS hazard pictograms by a hazard degree (for instance, a warning is 1, and danger, 2)¹⁴. Also, the energy consumptions for HPLC and UPLC are ≤ 1.5 kWh per sample and < 0.1kWh per sample, respectively¹⁴. Besides, hazards and generation of wastes were taken into account.

Method comparison between HPLC and UPLC

Five batches (27 samples) of combined amoxicillin/potassium clavulanate tablets previously analyzed by HPLC were re-analyzed by the described UPLC method. For all analytes, the correlation between the methods was tested by a Passing-Bablok fit. It was chosen since the preliminary check of the data samples for compliance with the normal distribution showed that data samples of amoxicillin (measured by UPLC method) and potassium clavulanate (measured by HPLC) did not correspond to a normal distribution (p < 0.01). This limits the use of ordinary least squares method for regression analysis. Therefore, the non-parametric Passing-Bablok method was taken¹⁵⁾. Comparison procedure was performed by Microsoft Excel 2013.

Results and discussion

Verification of HPLC Pharmacopoeial method

The British Pharmacopoeia proposes to use the HPLC method for the assay of medicines containing

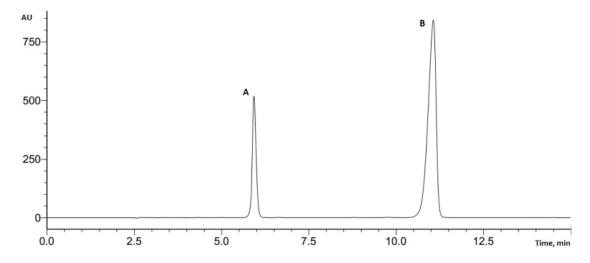


Fig. 2. Typical HPLC chromatogram (A – potassium clavulanate, B – amoxicillin)

potassium clavulanate and amoxicillin⁶⁾. For separation of components, a stainless steel column (250 mm \times 4.6 mm, with a particle size of 5 μm) packed with octadecylsilyl silica gel for chromatography is required. The mobile phase is a mixture of 5 volumes of methanol and 95 volumes of a 0.78% solution of sodium phosphate monobasic adjusted to pH 4.4 with phosphoric acid. The flow rate of the mobile phase should be 2 mL/min. The volume of injection is 20 μL , and the detection wavelength is set at 220 nm.

When the parameters on the HPLC system were applied, a high pressure was observed at this flow rate. Considering that, the flow rate was reduced to 1 mL/min. These changed parameters were used in further analysis of tablets containing potassium clavulanate and amoxicillin (Fig. 2).

Transfer of HPLC method to UPLC

Initially, the flow rate was reduced to 0.2 mL/min, and a smaller volume of the injection was applied (1 μ L). The analysis showed no separation between

the components; therefore, the flow rate was changed to $0.1\ mL/min$.

After reducing the flow rate, the coefficient of separation between components was 2.02 but still, the peak of potassium clavulanate was close to the void volume.

To increase separation, different buffer pH was tried in the range of 4.0–6.0. Changed conditions had no improvements for compounds separation, thus the buffer with 4.4 pH was left.

In the next step, we changed the ratio between buffer and methanol to 97:3 and reduced the volume of injection to 0.5 μ L. As a result, the separation between peaks became greater. Moreover, retention time for the peak of potassium clavulanate increased significantly and moved from the void volume.

And at the final step, we changed the ratio between buffer and methanol to 98:2 and set a volume of injection at $0.2~\mu L$.

These conditions were used in the assay of potassium clavulanate and amoxicillin in tablets (Fig.3).

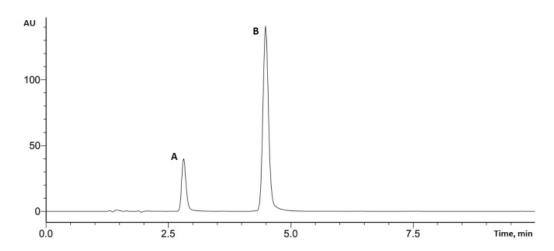


Fig. 2. Typical HPLC chromatogram (A – potassium clavulanate, B – amoxicillin)

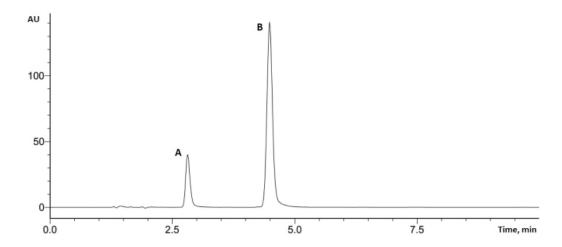


Fig. 3. Typical UPLC chromatogram (A – potassium clavulanate, B – amoxicillin)

Suitability of chromatographic methods

The main purpose of the system suitability test is to prove the reproducibility of the chromatographic system and to examine applied procedures, equipment, reagents, etc. The British Pharmacopoeia requires to control two parameters: the coefficient of asymmetry for potassium clavulanate and the coefficient of separation between potassium clavulanate and amoxicillin. Both methods have met all requirements, thus they can be recommended for applying in the analysis of medicines containing amoxicillin and potassium clavulanate (Table 1).

Table 1. Parameters of system suitability for both HPLC and UPLC methods

Parameter	Requirements	Found by HPLC	Found by UPLC
Coefficient of asymmetry for the peak of potassium clavulanate	At most 1.5	1.04	1.27
Coefficient of separation between the peaks of potassium clavulanate and amoxicillin	At least 3.5	15.46	7.93

Table 2. Linearity studies for HPLC and UPLC methods

Compound	Calibration curve	Correlation coefficient	Linear range, μg/mL	RSD, %	LOD, μg/mL	LOQ, μg/mL			
HPLC	HPLC								
Potassium clavulanate	y = 1.0016x - 0.1851	0.9999	320–480	0.31	0.8	2.4			
Amoxicillin	y = 1.007x - 0.7512	0.9999	400–600	0.14	1	3.3			
UPLC									
Potassium clavulanate	y = 0.9961x + 0.3857	0.9999	320–480	0.29	2	6.6			
Amoxicillin	y = 0.9981x + 0.1645	0.9999	400–600	0.16	3	10			

Table 3. The precision studies for potassium clavulanate and amoxicillin

Mothod	Commonant	Concentration,	Int	ra-day (n =	= 6)	Inter-day (n = 12)			
Method	Component	μg/mL	Found, %	RSD	Error	Found, %	RSD	Error*	
HPLC	Potassium clavulanate	80	101.86	0.23	0.34	101.98	0.43	0.30	
HPLC	Amoxicillin	100	98.98	0.31	0.43	99.10	0.37	0.49	
UPLC	Potassium clavulanate	80	102.43	0.28	0.31	102.73	0.57	0.43	
UPLC	Amoxicillin	100	99.10	0.26	0.27	98.85	0.52	0.39	

^{*}According to the ICH Guidelines it indicates systematic error of the calculated result introduced by the analytical method from its theoretical true value.

Table 4. The accuracy data of analysis by both procedures

	Amount measured (%)				Relative standard deviation (%)				Recovery (%)			
Reference value, %	HP	HPLC UP		LC HPLC		LC	UPLC		HPLC		UPLC	
	PC	AC	PC	AC	PC	AC	PC	AC	PC	AC	PC	AC
80	79.73	79.71	80.06	79.99	0.18	0.32	0.17	0.23	99.67	99.64	100.07	99.99
85	85.15	84.97	85.15	85.12	0.13	0.18	0.16	0.28	100.17	99.96	100.18	100.15
90	89.75	89.92	89.74	89.94	0.27	0.19	0.19	0.24	99.72	99.8.	99.71	99.93
95	95.12	94.88	95.05	95.10	0.21	0.16	0.17	0.11	100.13	99.87	100.05	100.11
100	100.06	99.97	100.13	99.87	0.19	0.19	0.16	0.16	100.06	100.08	100.13	99.87
105	105.09	105.08	105.15	104.63	0.18	0.1	0.17	0.17	100.09	100.09	100.14	99.64
110	110.03	110.21	109.92	110.09	0.24	0.23	0.22	0.15	100.03	100.19	99.93	100.08
115	114.95	114.81	114.87	115.02	0.21	0.12	0.23	0.12	99.96	99.83	99.88	100.02
120	119.87	120.11	119.86	119.96	0.29	0.20	0.30	0.27	99.89	100.09	99.88	99.97

Validation

Specificity

Specificity studies were performed for both HPLC and UPLC methods by analyzing blank, placebo (containing all components except potassium clavulanate and amoxicillin), standard, and test solutions. According to the obtained results, no interference in the analysed components quantification from blank or placebo solutions was found. Thus, these procedures can be applied in the analysis of tablets contain potassium clavulanate and amoxicillin.

Linearity, LOD, LOQ

The assay procedure should be linear within the range of application, which should overlap the possible values of concentrations of the active substance. Linearity studies were performed in the range of 80–120% (step – 5%). For this purpose, 9 model solutions were prepared, the concentration of which varied uniformly within the range of application (Table 2).

The LOD and LOQ were calculated by the following formulas:

$$LOD = \frac{3.3 \cdot \sigma}{S} \wedge LOQ = \frac{10 \cdot \sigma}{S},$$

where σ is the standard deviation of the response, and S is the slope of the calibration curve^{9, 10)}.

The values of LOD and LOQ are presented in Table 2.

Precision

The precision study was performed within 2 days by different analysts. Test solution contained 100% of potassium clavulanate and amoxicillin. The results of the studies are shown in Table 3.

Stability

Stability studies have been performed for both procedures within 24 hours for a standard solution. It was found that the solutions were stable. Deviations of analyte peaks areas in the HPLC method for potassium clavulanate and amoxicillin were 0.338% and 0.362%, respectively. In the case of UPLC, potassium clavulanate and amoxicillin showed deviations of 0.391% and 0.164%, respectively.

Accuracy

To test accuracy within the range of analytical procedure, 9 test solutions with a known amount of potassium clavulanate (PC) and amoxicillin (AC) were prepared, following all stages of the analytical procedure. The measured solutions were in the range of 80–120% (Table 4). Both procedures have satisfactory results, thus they can be used in the analysis of medicines.

Eco-scale calculation

Both methods for simultaneous assay of amoxicillin and potassium clavulanate were assessed by their "greenness" using the Eco-scale to find a method with less environmental impact. The UPLC method showed a slightly higher Eco-scale value than the HPLC. The total scores for HPLC and UPLC were 82 and 85 respectively (Table 5).

Method comparison between HPLC and UPLC

The methods comparison was performed using the values obtained in the quantitative assessment of 27 tablets samples, measured by both HPLC and UPLC methods. The assessment was done using Passing-Bablok regression equations (estimation of α , β), 95% CI for a, b (test of hypothesis $\beta = 1$ and $\alpha = 0$), a test of linearity assumption (Table 6) and corresponding

Table 5. Assessment of the HPLC and UPLC methods

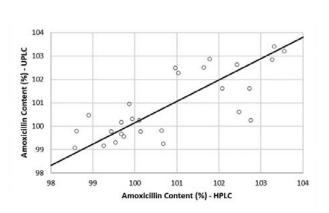
	Penalt	Penalty points		
	HPLC	UPLC		
Reagents				
Methanol 0.75 mL	6	6		
Water 14.25 mL	0	0		
Sodium phosphate monobasic	0	0		
Instrument				
HPLC	1	0		
Waste	8	6		
Occupation hazard	3	3		
Sample preparation				
Water 200 mL	0	0		
Total penalty points	18	15		
Analytical Eco-scale total score	82	85		

regression plots with real measurement data (Fig. 4)¹⁵). In plots, the regression lines (continuous) are in the range of amoxicillin and potassium clavulanate estimated quantitative contents from 98 to 104%. The real measurement data are presented by dots. The statistical test of the linearity assumption proved linearity between the HPLC and UPLC methods for both amoxicillin and potassium clavulanate (Table 6). The 95% CI included the value 1 for the slope and the value 0 for the intercept for both studied substances, which demonstrated that there is no statistically significant difference between the old and the new method.

Conclusions

The HPLC procedure for the simultaneous assay of amoxicillin and potassium clavulanate was successfully transferred to the UPLC. To optimize the conditions of the analysis, we have changed the ratio of mobile phase components, flow rate, and injection volume. Furthermore, the run time was shortened from 15 min to 7.5 min. The HPLC procedure was verified to further implementation into the SPhU monograph, and the UPLC procedure was fully validated. Both methods have met all requirements, and could be recommended for applying in the analysis of medicines containing amoxicillin and potassium clavulanate. The methods were compared in terms of environmental friendliness. In general, both methods have shown themselves to be ecofriendly, although the UPLC method showed a slightly better result. A Passing-Bablok regression method comparison showed the similarity of methods in obtained results. Both methods are reliable and could be used in laboratories during the quality control process.

Conflict of interest: none.



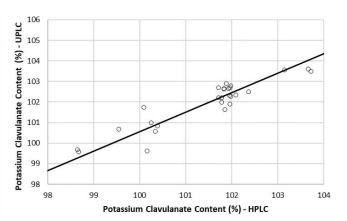


Fig. 4. Regression plots of the comparison between HPLC and UPLC for amoxicillin (a) and potassium clavulanate (a) by the Passing-Bablok method

Table 6. Testing the equality of measurements from HPLC and UPLC methods for simultaneous assay of amoxicillin and potassium clavulanate

Estimated Param Y = bx+a	neter	Amoxicillin	Potassium Clavulanate		
	В	0.91	0.95		
Slone R	b-low (Low 95% CI)	0.66	0.79		
Slope β	b-up (Up 95% CI)	1.25	1.26		
	Test of the hypothesis $\beta = 1$	+	+		
Intercept α	A	8.73	5.73		
	a-low (Low 95% CI)	-25.27	-26.30		
	a-up (Up 95% CI)	34.21	21.95		
	Test of the hypothesis $\alpha = 0$	+	+		
	h-stat	0.53	1.06		
Statistical test of the linearity	p-level	0.93	0.20		
		+	+		
assumption	p-level > 0.05	No significant deviation from linearity			

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